

# GAS CHROMATOGRAPHY

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# CLIL Module Plan:

<b>School</b>	ITET Mantegna - Mantova
<b>School year</b>	5°
<b>Subject</b>	Analisi Chimica e Strumentale
<b>Topic</b>	Gas Chromatography
<b>CLIL Language</b>	English
<b>Module lenght:</b> 16 hours	<b>Unit 1 – 3 hours</b> Introduction to Gas Chromatography (GC); how GC works; the chromatogram. A brief mention of Gas Chromatography Mass Spectrometry (GC-MS); block diagram of the instrument
	<b>Unit 2 – 5 hours</b> GC fundamentals: type of columns (packed vs. capillary), stationary phase selectivity and mechanism of GC separation
	<b>Unit 3 – 4 hours</b> Van Deemter equation: Eddy diffusion, longitudinal diffusion mass transfer; the concept of the theoretical plate and the Van Deemter polt
	<b>Unit 4 – 3 hours</b> Split/Splitless mode and FID
	<b>Unit 5 – 1 hour</b> Online summative assessment

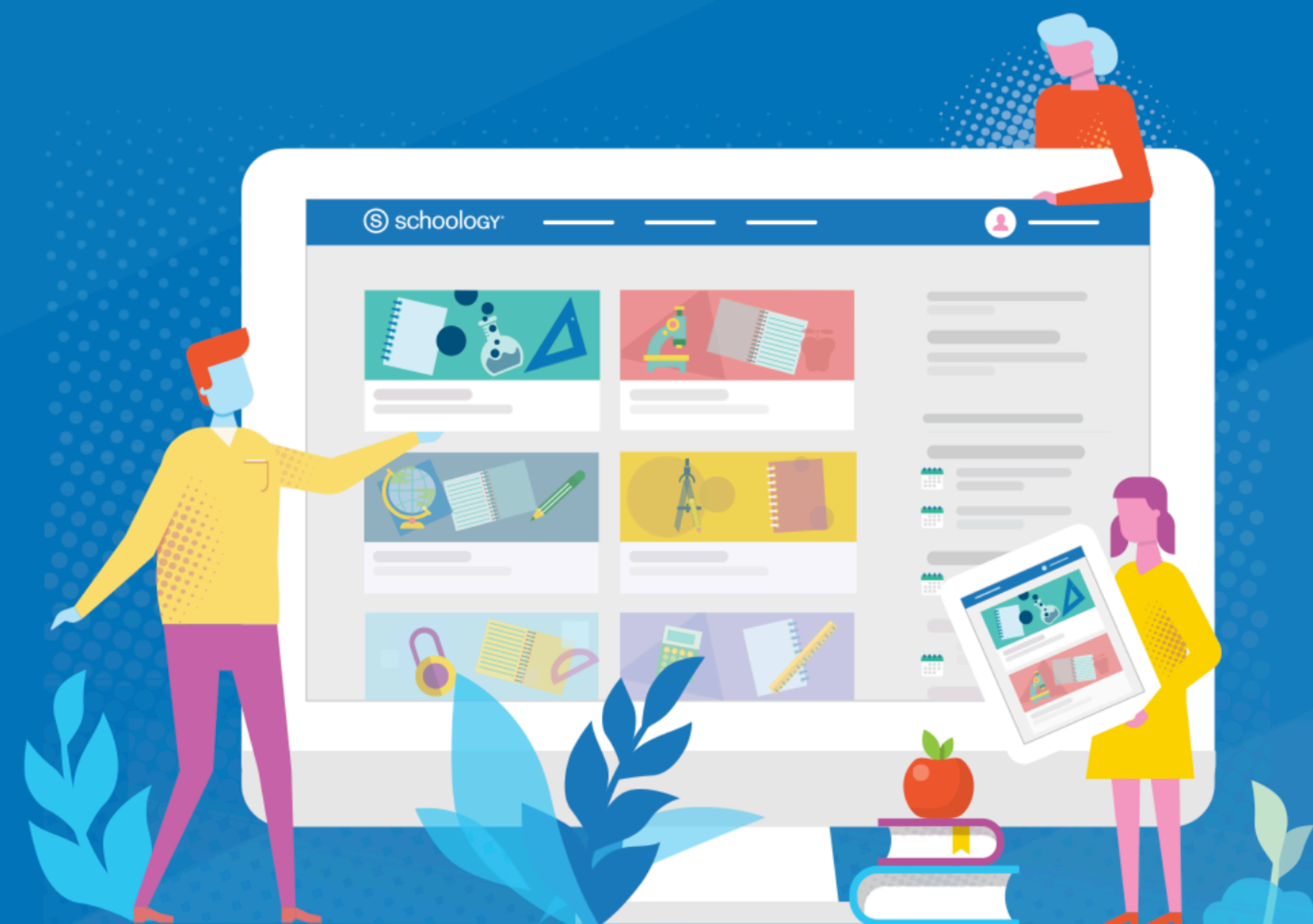


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## Gas Chromatography. How GC instrument works. The Chromatogram

Due: Friday, December 18, 2020 at 8:00 am

An introductory video from the Royal Society of Chemistry on Gas Chromatography using a flame ionisation detector (FID) with a brief mention of Gas Chromatography Mass Spectrometry (GCMS).

Follow steps 1 - 4:

1- watch the video once (no more than 10 min) focusing on the words related to the GC instrument sections;

2- complete the quiz;

3 - draw a block diagram of a GC and upload it on the Discussion section (no more than 5 min); then add a post to one classmates' block diagram (no more than 5 min);

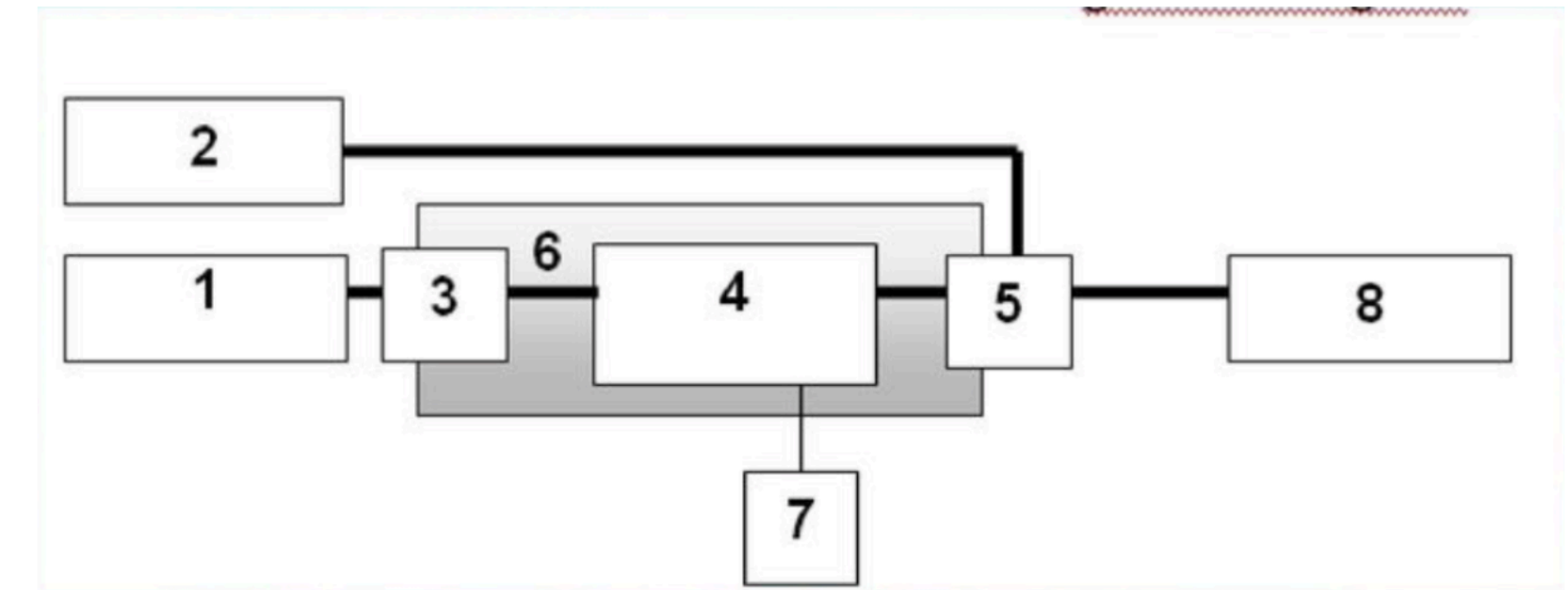
4 - on your own rewatch the video, take notes, reconstruct the text and upload it on the Assignment section (no more than 25 min).



## GC BLOCK DIAGRAM

· Due Tuesday, December 8, 2020 at 11:59 pm

### 1 GC BLOCK DIAGRAM



1. \_
2. \_
3. \_
4. \_
5. \_
6. \_
7. \_
8. \_

Fill in the Blank - 8 points



# How separation occurs in GC column

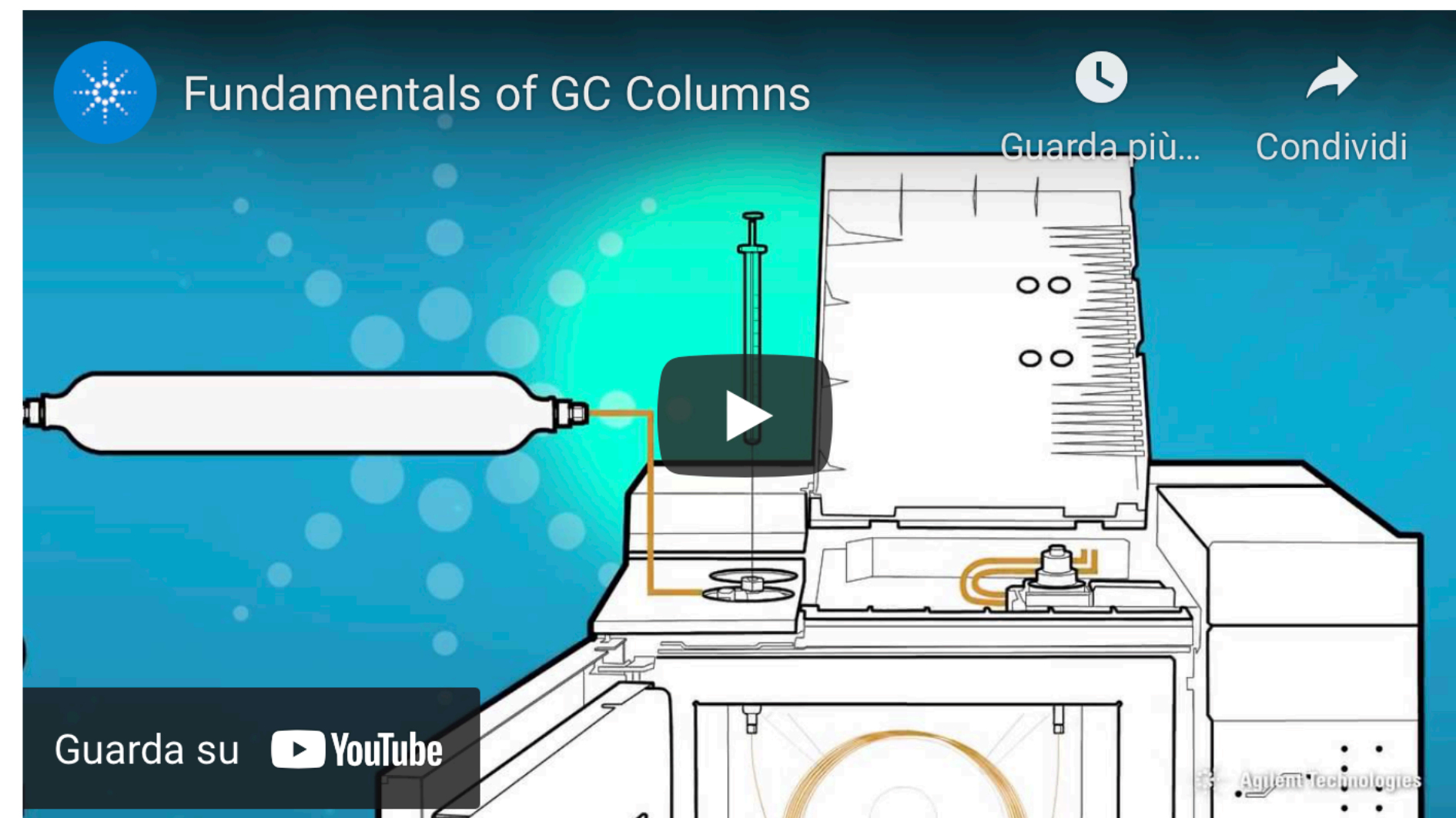
Chimica Analitica e Strumentale: 5AA

 CLIL - Gas Chromatography

## Fundamentals of GC

In this video you will explore the theory of GC columns and the separations. The objectives follow:

1. identify packed vs. capillary column
2. describe the features and differences of a packed vs. capillary column
3. describe how peaks separation occurs in a GC column
4. apply the concept of the theoretical plate and the Van Deemter plot to the GC method set-up

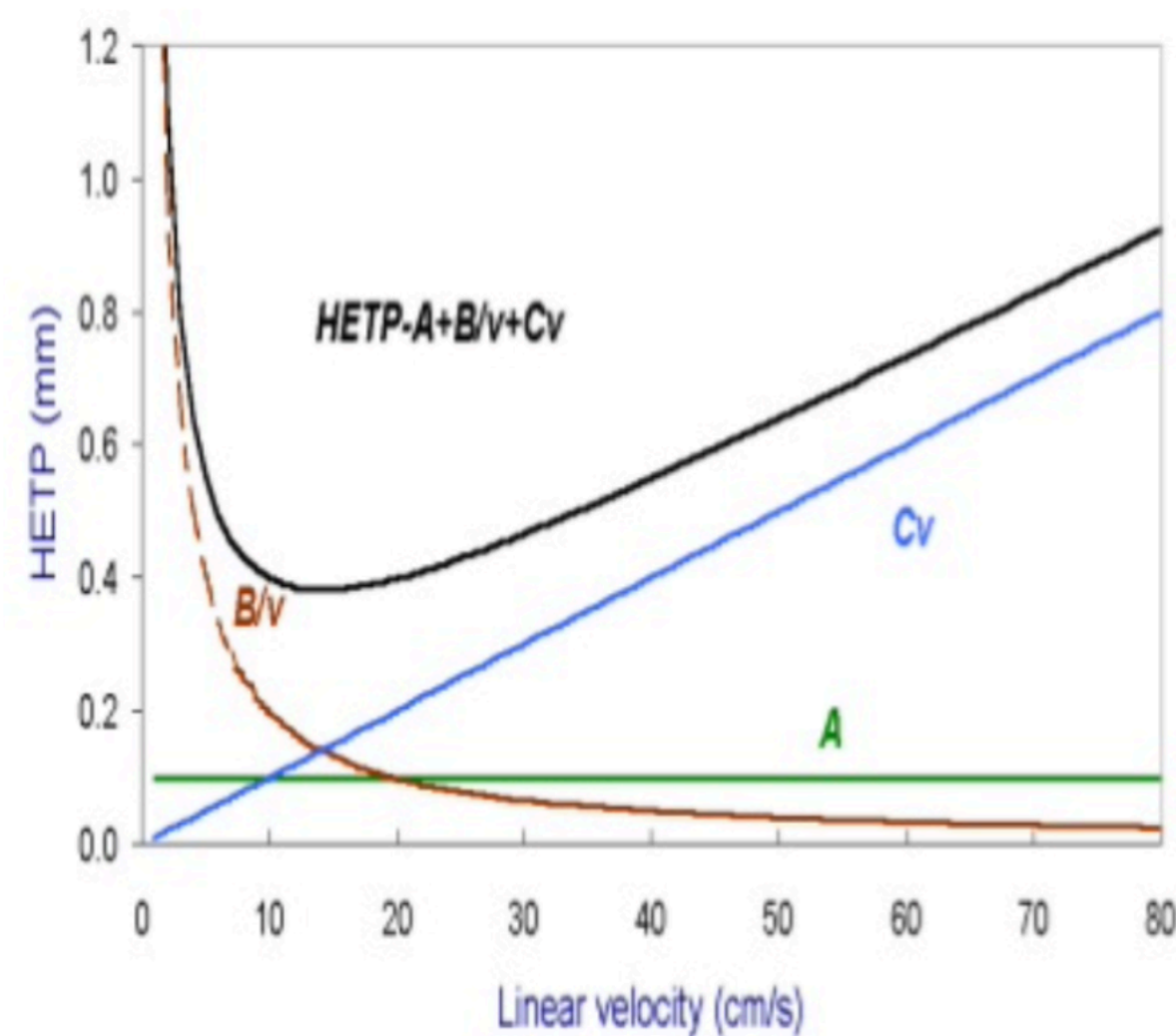


Students used Edpuzzle  
to  
cut from Fundamentals  
of GC video  
the section of interest.



Visual Dictation:

Van Deemter Plot

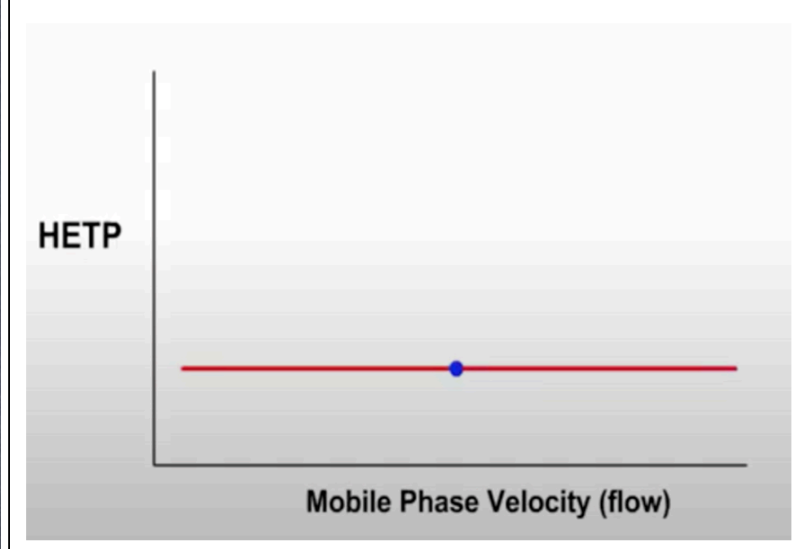


The term 'A' is independent of flow rate of the mobile phase

The term B/u decreases drastically in the beginning with increase in the flow rate of mobile phase. Increase in the flow rate beyond particular value, leads to slow decrease in the value of B/u.

The term Cu increases with increase in the flow rate

- Posted Thu Dec 31, 2020 at 7:44 pm
- Eddy diffusion term
- Longitudinal diffusion
- Mass transfer
- HETP vs flow velocity - Eddy diffusion
- HETP vs. flow velocity - longitudinal diffusion
- HETP vs. flow velocity - mass transfer

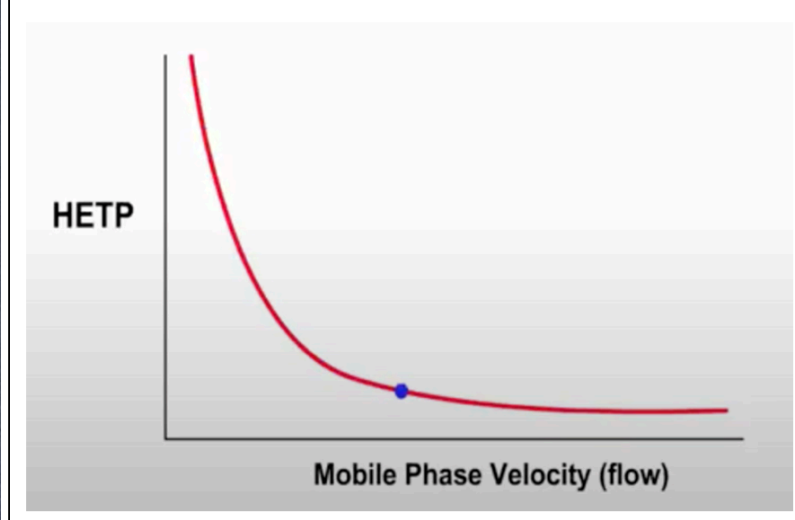


Answer the questions

1. What is the name of the process displayed and which is the corresponding term in Van Deemter equation?

2. Give a brief description of the process.

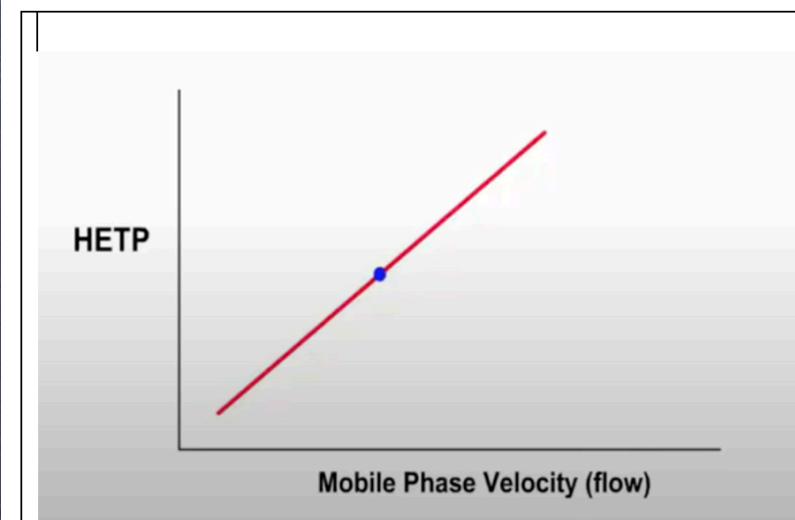
3. Does it have any effect in GC separation? In which way?



1. What is the name of the process displayed which is the corresponding term in Van Deemter equation?

2. Give a brief description of the process.

3. Does it have any effect in GC separation? In which way?



1. What is the name of the process displayed and which is the corresponding term in Van Deemter equation?

2. Give a brief description of the process.

3. Does it have any effect in GC separation? In which way?

Plot the overall Van Deemter graph

1. Write the Van Deemter equation.

2. How is the average linear velocity defined?

3. Describe the Van Deemter plot and Explain how is it used for?



# Self assessment test:



SPLIT/SPLITLESS MODE and FID



GC test

· Due Friday, February 26, 2021 at 11:59 pm

